

Supporting Information for:

## Photolysis of $\alpha$ -Azidoacetophenones: Trapping of Triplet Alkyl Nitrenes in Solution

Sarah Mandel, Jeanette A. Krause Bauer and Anna D. Gudmundsdottir\*

Department of Chemistry, University of Cincinnati, Cincinnati, OH 45221-0172

### $\alpha$ -azidoacetophenone (1a)

IR (KBr,  $\nu_{\max}$ ): 2104, 1695, 1217  $\text{cm}^{-1}$ .

$^1\text{H-NMR}$  (250 MHz,  $\text{CDCl}_3$ )  $\delta$  7.9 (s, 7Hz, 2H), 7.6 (t, 7Hz, 1H), 7.5 (t, 7Hz, 2H), 4.5 (s, 2H) ppm.

MS  $m/z$  (rel. intensity): 161( $\text{M}^+$ , 1) 105 (100), 77 (80).

### 2-benzoylamino-1-phenylethanone (3a)

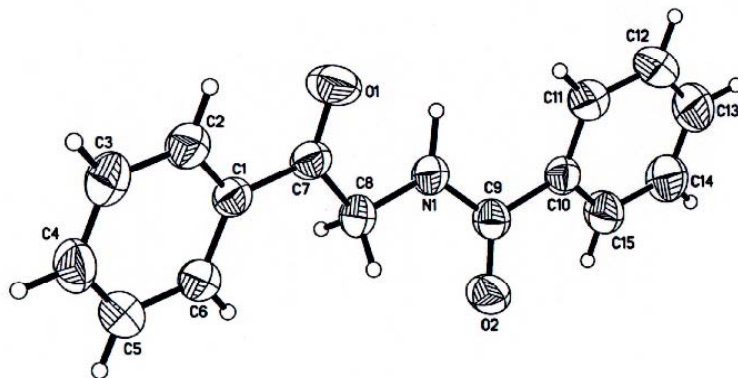
MP: 122-124°C (lit.<sup>1</sup>, 123-125°C).

IR (KBr,  $\nu_{\max}$ ): 3358, 1692, 1636, 1523, 1365, 1224  $\text{cm}^{-1}$ .

$^1\text{H-NMR}$  (250 MHz,  $\text{CDCl}_3$ )  $\delta$  8.04 (d, 7Hz, 2H), 7.89 (d, 7 Hz, 2H), 7.66 (t, 7Hz, 1H), 7.5 (m, 5H), 7.3 (broad s, 1H), 4.98 (d, 4Hz, 2H) ppm.

MS  $m/z$  (rel. intensity): 239 ( $\text{M}^+$ , 10), 105 (100).

### CRYSTAL DATA OF 3a



Chemical formula:  $\text{C}_{15}\text{H}_{13}\text{NO}_2$

Crystal system: Monoclinic

Space group:  $\text{P2}_1/\text{c}$

<sup>1</sup> Demeuov, N. B.; Akhmedzhanova, V. I.; Moldagulov, M. A.; Shakirov, R. S.; *Chem. Nat. Compd. (Engl. Transl.)*, **1998**, 34, 484.

$a = 13.063(2) \text{ \AA}$        $\alpha = 90^\circ$   
 $b = 5.3745(6) \text{ \AA}$        $\beta = 99.449(2)^\circ$   
 $c = 17.450(2) \text{ \AA}$        $\gamma = 90^\circ$   
Volume =  $1208.5(2) \text{ \AA}^3$        $Z = 4$   
 $\rho_{\text{(calcd)}} = 1.315 \text{ Mg/m}^3$        $\mu = 0.088 \text{ mm}^{-1}$

No. of reflections for lattice parameters: 1707  
 $\theta$  range for lattice parameters ( $^\circ$ ): 2.37 to 28.30

## EXPERIMENTAL

### Crystallization

Crystallization source: diethyl ether  
Crystal size (mm): 0.34 x 0.14 x 0.08  
Crystal description: colorless rod

### Data Collection

Temperature (K): 296(2)  
Mo  $K\alpha$  Radiation:  $\lambda = 0.71073 \text{ \AA}$   
 $\theta$  range for data collection ( $^\circ$ ): 2.37 to 28.30  
No. of reflections collected: 7654  
No. of independent reflections: 2984  
 $R_{\text{int}} = 0.0629$   
Absorption correction (Tmin, Tmax): 0.971, 0.993

## REFINEMENT

Full-matrix least-squares on  $F^2$   
No. of reflections  $I > 2\sigma(I)$ : 1551  
Final R indices [ $I > 2\sigma(I)$ ]:  $R1 = 0.0582$ ,  $wR2 = 0.1213$   
R indices (all data):  $R1 = 0.1252$ ,  $wR2 = 0.1489$   
Extinction coefficient =  $0.002(2)$   
 $S = 1.000$   
Largest residual peak:  $0.178 \text{ e\AA}^{-3}$

### Treatment of Hydrogen Atoms

The hydrogen atom on N1 was found directly and held fixed in subsequent refinements.  
The remaining hydrogen atom positions were calculated based on geometrical criteria and treated with a riding model. The isotropic temperature factors for the H-atoms were defined as  $1.2U_{\text{eq}}$  of the adjacent atom.

Table 1. Atomic coordinates [ $\times 10^4$ ] and equivalent isotropic displacement parameters [ $\text{\AA}^2 \times 10^3$ ]. U(eq) is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

|       | x        | y       | z       | U(eq) |
|-------|----------|---------|---------|-------|
| O(1)  | -420(1)  | 1725(3) | 3864(1) | 69(1) |
| O(2)  | -1785(1) | 8715(3) | 2407(1) | 60(1) |
| N(1)  | -1352(1) | 4783(3) | 2770(1) | 47(1) |
| C(1)  | 988(1)   | 4299(4) | 4390(1) | 40(1) |
| C(2)  | 1291(2)  | 2709(4) | 5012(1) | 48(1) |
| C(3)  | 2164(2)  | 3207(4) | 5548(1) | 55(1) |
| C(4)  | 2747(2)  | 5288(5) | 5466(1) | 58(1) |
| C(5)  | 2465(2)  | 6877(4) | 4851(1) | 58(1) |
| C(6)  | 1582(2)  | 6407(4) | 4316(1) | 49(1) |
| C(7)  | 33(2)    | 3690(4) | 3829(1) | 43(1) |
| C(8)  | -357(1)  | 5551(4) | 3210(1) | 46(1) |
| C(9)  | -2015(2) | 6483(4) | 2412(1) | 44(1) |
| C(10) | -3051(1) | 5614(4) | 2005(1) | 43(1) |
| C(11) | -3542(2) | 3498(4) | 2221(1) | 57(1) |
| C(12) | -4521(2) | 2885(5) | 1831(2) | 73(1) |
| C(13) | -4998(2) | 4331(5) | 1228(2) | 75(1) |
| C(14) | -4511(2) | 6404(5) | 1006(1) | 68(1) |
| C(15) | -3549(2) | 7052(5) | 1397(1) | 55(1) |

Table 2. Bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ].

|             |          |             |          |
|-------------|----------|-------------|----------|
| O(1)-C(7)   | 1.217(2) | O(2)-C(9)   | 1.237(2) |
| N(1)-C(9)   | 1.340(2) | N(1)-C(8)   | 1.457(2) |
| C(1)-C(2)   | 1.388(3) | C(1)-C(6)   | 1.391(3) |
| C(1)-C(7)   | 1.489(3) | C(2)-C(3)   | 1.376(3) |
| C(3)-C(4)   | 1.374(3) | C(4)-C(5)   | 1.374(3) |
| C(5)-C(6)   | 1.383(3) | C(7)-C(8)   | 1.499(3) |
| C(9)-C(10)  | 1.497(3) | C(10)-C(15) | 1.386(3) |
| C(10)-C(11) | 1.387(3) | C(11)-C(12) | 1.387(3) |
| C(12)-C(13) | 1.373(4) | C(13)-C(14) | 1.370(4) |
| C(14)-C(15) | 1.372(3) |             |          |

|                   |          |                   |          |
|-------------------|----------|-------------------|----------|
| C(9)-N(1)-C(8)    | 120.3(2) | C(2)-C(1)-C(6)    | 118.8(2) |
| C(2)-C(1)-C(7)    | 118.6(2) | C(6)-C(1)-C(7)    | 122.5(2) |
| C(3)-C(2)-C(1)    | 120.7(2) | C(4)-C(3)-C(2)    | 119.9(2) |
| C(3)-C(4)-C(5)    | 120.3(2) | C(4)-C(5)-C(6)    | 120.1(2) |
| C(5)-C(6)-C(1)    | 120.1(2) | O(1)-C(7)-C(1)    | 121.3(2) |
| O(1)-C(7)-C(8)    | 120.3(2) | C(1)-C(7)-C(8)    | 118.4(2) |
| N(1)-C(8)-C(7)    | 111.1(2) | O(2)-C(9)-N(1)    | 121.7(2) |
| O(2)-C(9)-C(10)   | 120.1(2) | N(1)-C(9)-C(10)   | 118.2(2) |
| C(15)-C(10)-C(11) | 119.0(2) | C(15)-C(10)-C(9)  | 117.6(2) |
| C(11)-C(10)-C(9)  | 123.3(2) | C(12)-C(11)-C(10) | 119.5(2) |
| C(13)-C(12)-C(11) | 120.5(2) | C(14)-C(13)-C(12) | 120.3(2) |
| C(13)-C(14)-C(15) | 119.7(2) | C(14)-C(15)-C(10) | 121.0(2) |

Table 3. Anisotropic displacement parameters [ $\text{\AA}^2 \times 10^3$ ]. The anisotropic displacement factor exponent takes the form:  $-2\pi^2[(ha^*)^2U_{11} + \dots + 2hka^*b^*U_{12}]$

|       | U11   | U22   | U33    | U23    | U13    | U12    |
|-------|-------|-------|--------|--------|--------|--------|
| O(1)  | 74(1) | 45(1) | 77(1)  | 7(1)   | -21(1) | -16(1) |
| O(2)  | 54(1) | 48(1) | 69(1)  | 7(1)   | -11(1) | -6(1)  |
| N(1)  | 39(1) | 44(1) | 52(1)  | 0(1)   | -6(1)  | 2(1)   |
| C(1)  | 39(1) | 36(1) | 42(1)  | -3(1)  | 1(1)   | 3(1)   |
| C(2)  | 47(1) | 46(1) | 48(1)  | 3(1)   | 2(1)   | 4(1)   |
| C(3)  | 51(1) | 61(2) | 50(1)  | 4(1)   | 0(1)   | 11(1)  |
| C(4)  | 44(1) | 65(2) | 59(1)  | -13(1) | -6(1)  | 5(1)   |
| C(5)  | 49(1) | 52(1) | 69(1)  | -5(1)  | 0(1)   | -5(1)  |
| C(6)  | 46(1) | 44(1) | 55(1)  | 3(1)   | 0(1)   | 0(1)   |
| C(7)  | 45(1) | 37(1) | 46(1)  | -3(1)  | 3(1)   | 3(1)   |
| C(8)  | 38(1) | 48(1) | 49(1)  | 3(1)   | -2(1)  | 3(1)   |
| C(9)  | 41(1) | 47(1) | 42(1)  | 0(1)   | 2(1)   | 0(1)   |
| C(10) | 38(1) | 43(1) | 48(1)  | -4(1)  | 3(1)   | 2(1)   |
| C(11) | 49(1) | 47(1) | 72(1)  | 3(1)   | 2(1)   | 0(1)   |
| C(12) | 55(2) | 54(2) | 107(2) | -4(2)  | 4(1)   | -14(1) |
| C(13) | 45(1) | 75(2) | 96(2)  | -16(2) | -14(1) | -4(1)  |
| C(14) | 52(1) | 72(2) | 72(2)  | 0(1)   | -16(1) | 4(1)   |
| C(15) | 47(1) | 57(2) | 58(1)  | 6(1)   | -4(1)  | -1(1)  |

Table 4. Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ).

|       | x        | y       | z       | U(eq) |
|-------|----------|---------|---------|-------|
| H(1)  | -1589    | 2920    | 2787    | 56    |
| H(2)  | 901(2)   | 1292(4) | 5068(1) | 57    |
| H(3)  | 2358(2)  | 2135(4) | 5965(1) | 66    |
| H(4)  | 3335(2)  | 5624(5) | 5829(1) | 69    |
| H(5)  | 2868(2)  | 8270(4) | 4794(1) | 69    |
| H(6)  | 1385(2)  | 7502(4) | 3905(1) | 59    |
| H(8A) | -434(1)  | 7156(4) | 3448(1) | 55    |
| H(8B) | 145(1)   | 5727(4) | 2862(1) | 55    |
| H(11) | -3216(2) | 2498(4) | 2623(1) | 69    |
| H(12) | -4858(2) | 1484(5) | 1979(2) | 88    |
| H(13) | -5654(2) | 3901(5) | 969(2)  | 90    |
| H(14) | -4831(2) | 7368(5) | 592(1)  | 82    |
| H(15) | -3226(2) | 8478(5) | 1251(1) | 66    |

Detailed X-ray crystallographic data (excluding structure factors) are available from the Cambridge Crystallographic Data Centre, as a supplementary publication no. CCDC-1557621. Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax (+44)-1223-336-033; e-mail [deposit@ccdc.cam.ac.uk](mailto:deposit@ccdc.cam.ac.uk)).

**$\alpha$ -azidoacetophenone (1b)**

MP: 78-81°C (lit. 86-87°C)<sup>2</sup>.

IR (KBr,  $\nu_{\max}$ ): 2105, 1690, 1587, 1219, 8181  $\text{cm}^{-1}$ .

<sup>1</sup>H-NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  7.7 (d, 8Hz, 2H), 7.6 (d, 8Hz, 2H), 4.53 (s, 2H) ppm.

MS m/z (rel. intensity): 240/238 ( $\text{M}^+$ , 30), 185/183 (100) 157/155 (75),

**2-(4-bromobenzoylamino)-1-(4-bromophenyl)ethanone (3b)**

MP: 194-196°C (lit. 193-197°C)<sup>3</sup>.

IR (KBr,  $\nu_{\max}$ ): 3333, 1689 1638, 1587, 1520, 12210, 994  $\text{cm}^{-1}$ .

<sup>1</sup>H-NMR (250 MHz,  $\text{CDCl}_3$ )  $\delta$  7.90 (d, 7 Hz, 2H), 7.7 (m, 6H), 7.20 (broad s, 1H), 4.91 (d, 4Hz, 2H) ppm.

MS m/z (rel. intensity): 399/397/395 ( $\text{M}^+$ , 2), 183/185 (100) 155/157 (50).

<sup>2</sup> Boyer, J.H.; Straw, D. J. Am. Chem. Soc. **1952**, 74, 4506,

<sup>3</sup> Dann, O.; *Justus Liebigs Ann. Chem.* **1975**; 160.